

NAME (Print): _____

SIGNATURE: _____

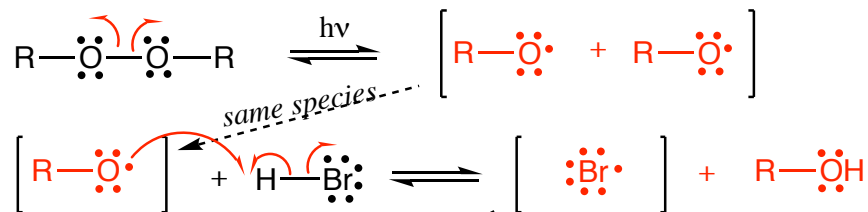
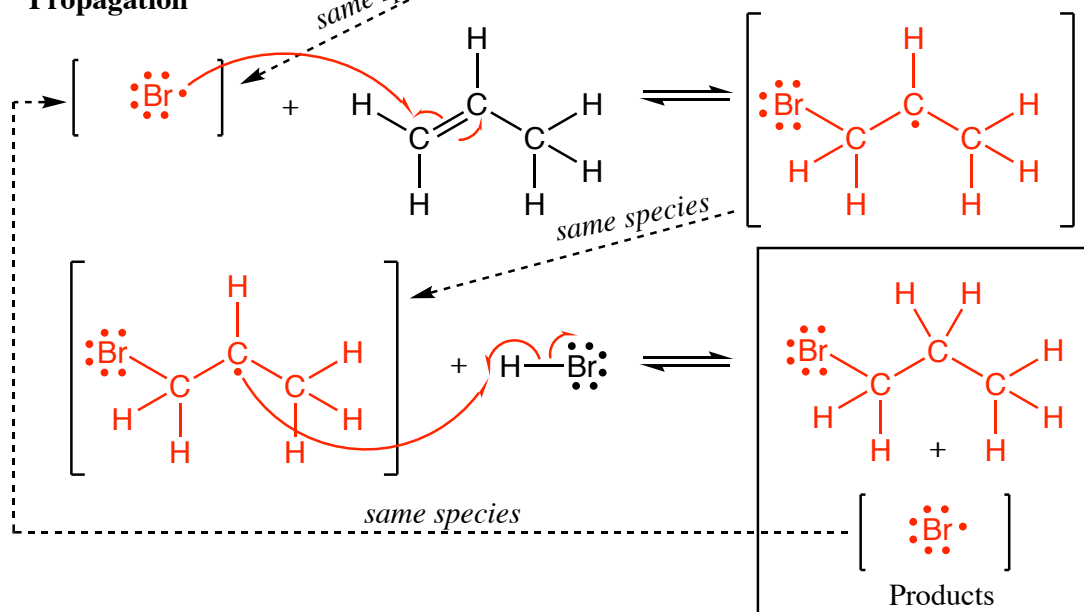
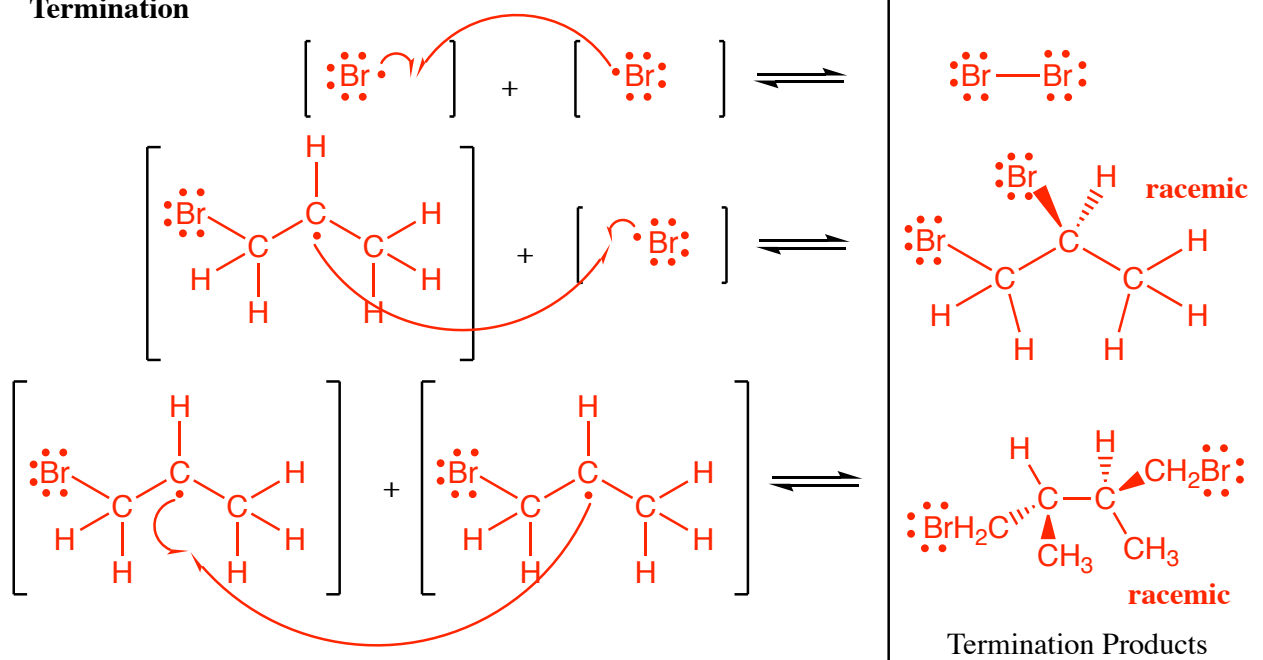
Chemistry 320M/328M
Dr. Brent Iverson
8th Homework
November 1, 2022

Please print the
first three letters
of your last name
in the three boxes

--	--	--

Score: _____

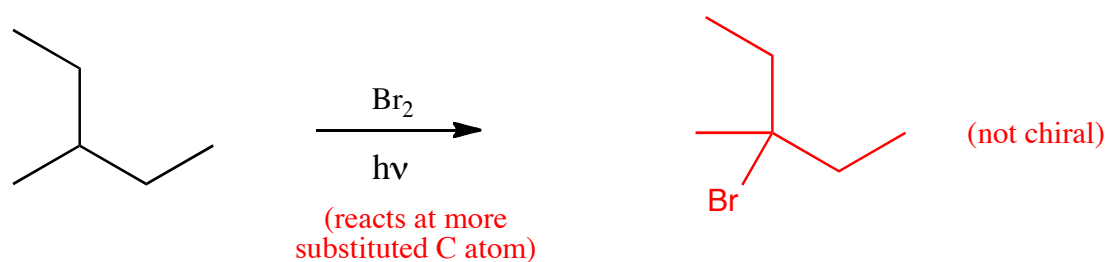
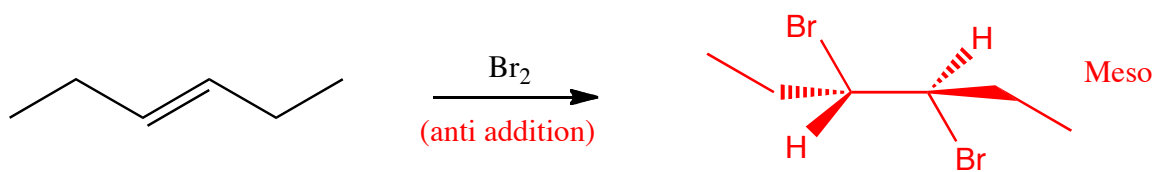
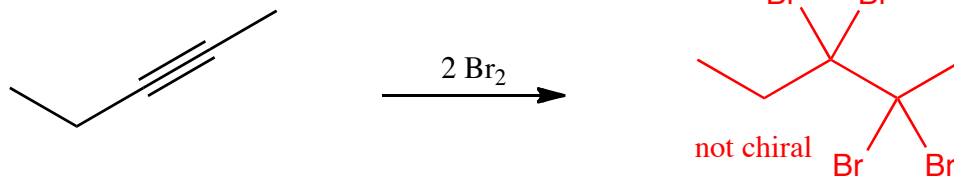
1. Complete the following initiation and propagation steps for the addition of HBr via a free radical chain reaction mechanism. Use appropriate arrows to show movement of electron density, and show all non-bonding electrons as dots and show any formal charges. **If any of the species are really a racemic mixtures of enantiomers, you only need to draw one stereoisomer and write "racemic"**

Initiation**Propagation****Termination**

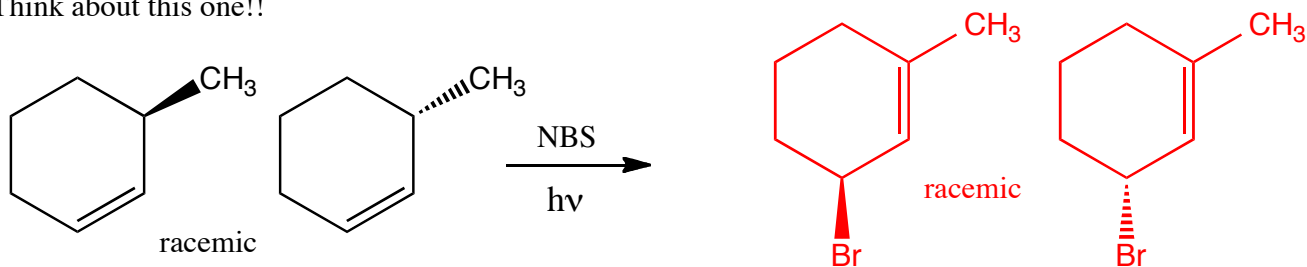
2. Succinctly define Hammond's postulate.

Hammond's postulate says that the transition state for an exothermic step resembles the reactants more than the products of that step, and the structure of the transition state for an endothermic step resembles the products more than the reactants of that step.

3. For the following, complete the reactions with the predominant product or products. You must indicate stereochemistry with wedges and dashes. If a racemic mixture is created, you must write "racemic" under the structures.

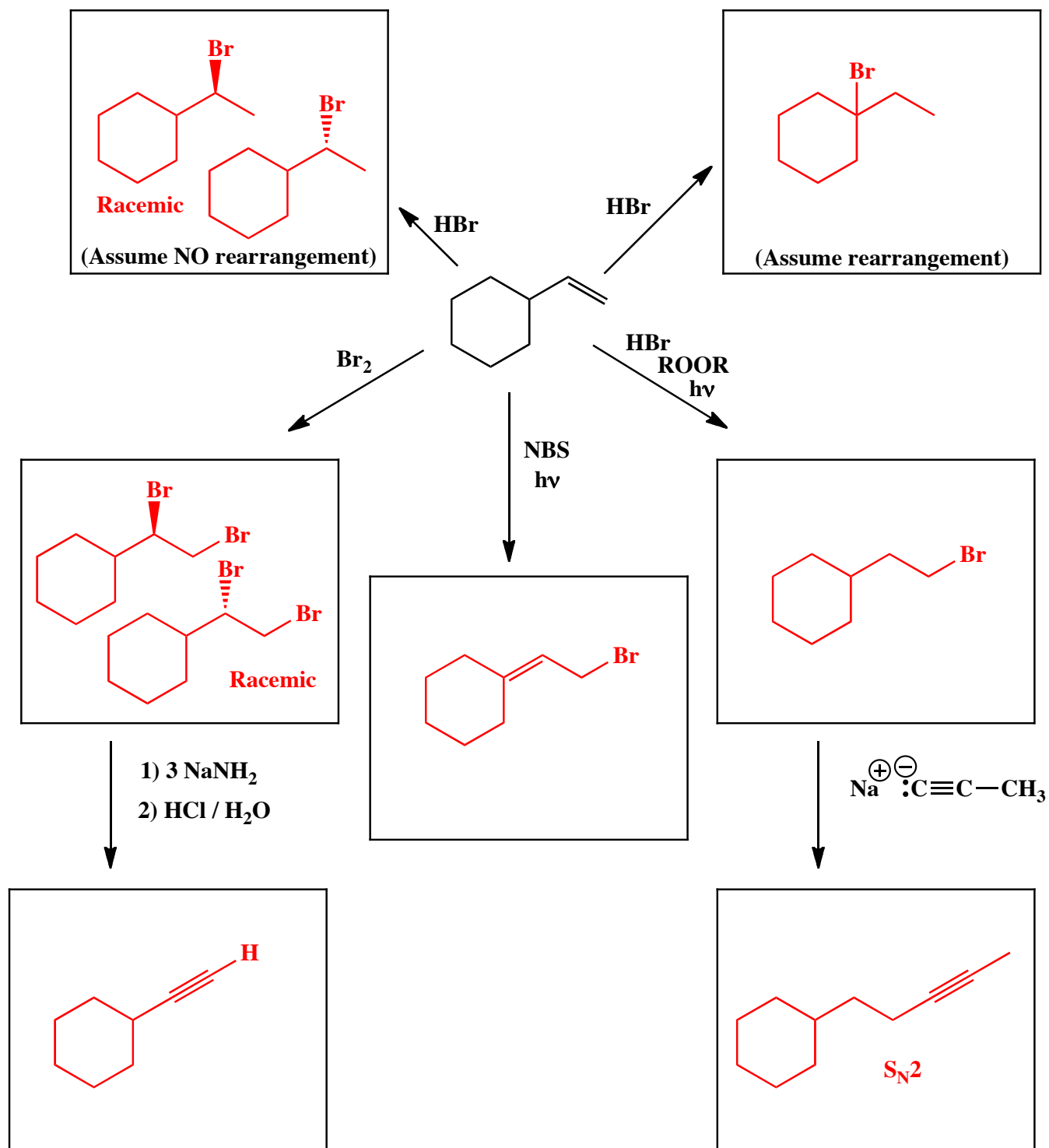


Think about this one!!

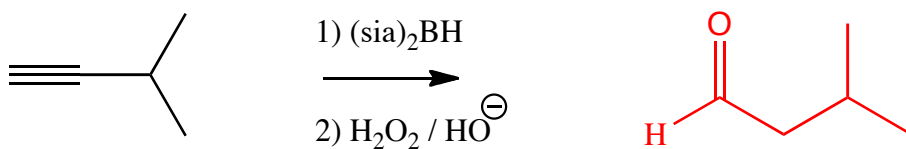
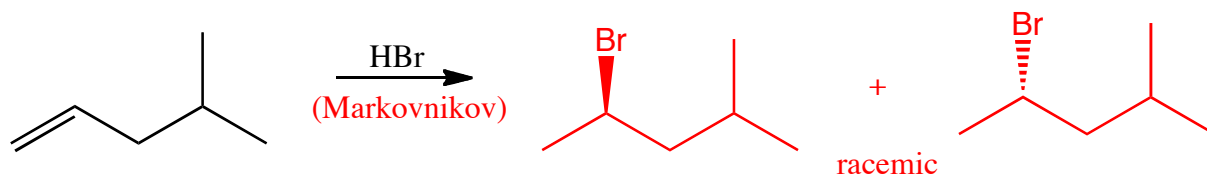
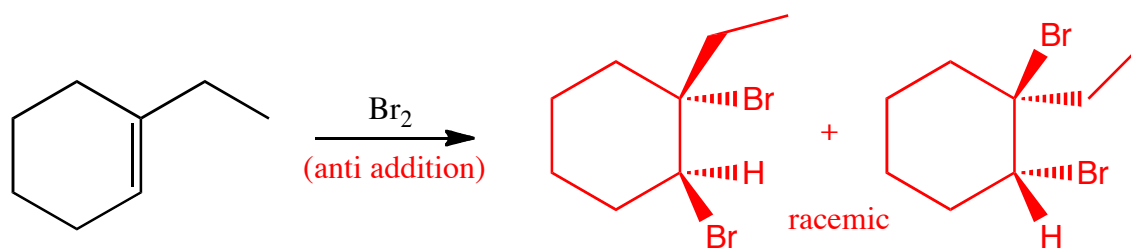
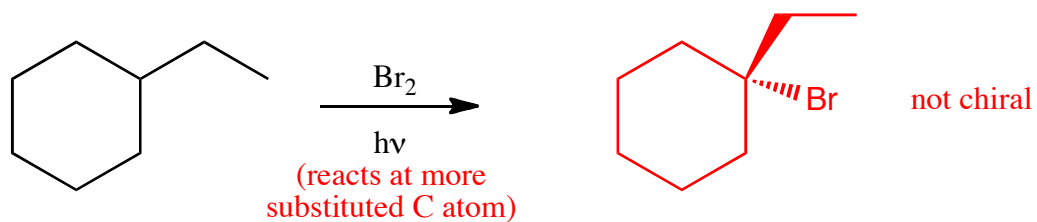


The most stable alkene removes the chiral center at methyl.

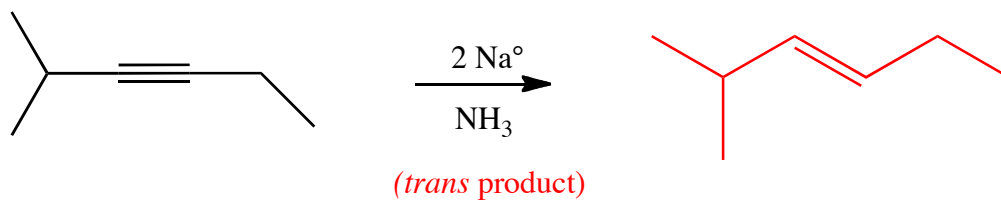
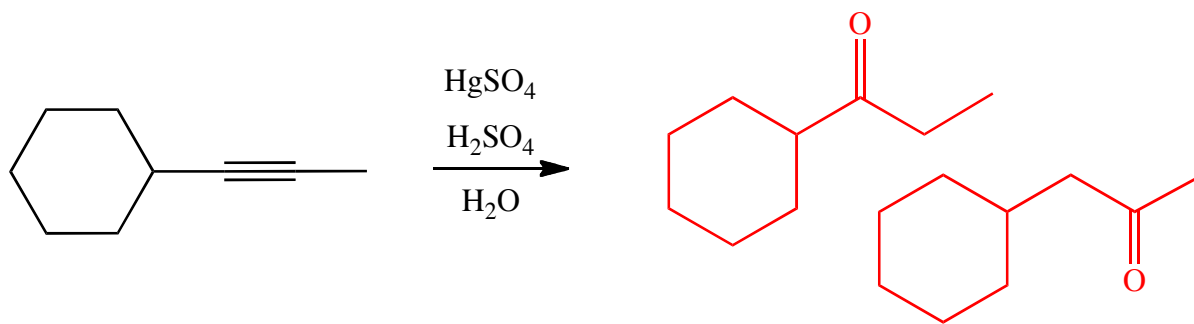
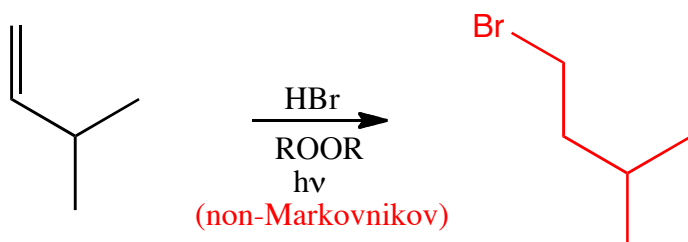
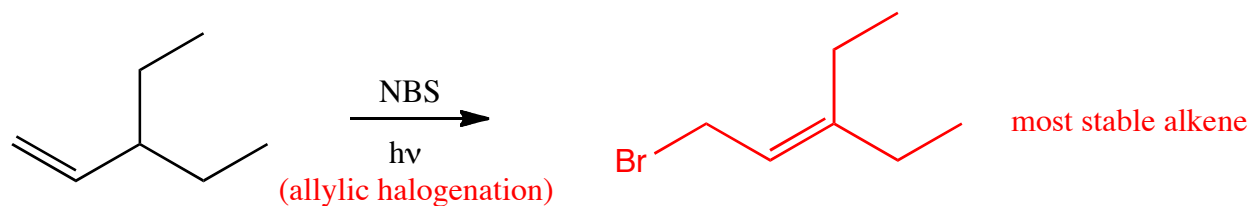
4. For the following, complete the reactions with the predominant product or products. You must indicate stereochemistry with wedges and dashes. If a racemic mixture is created, you must write "racemic" under the structures.



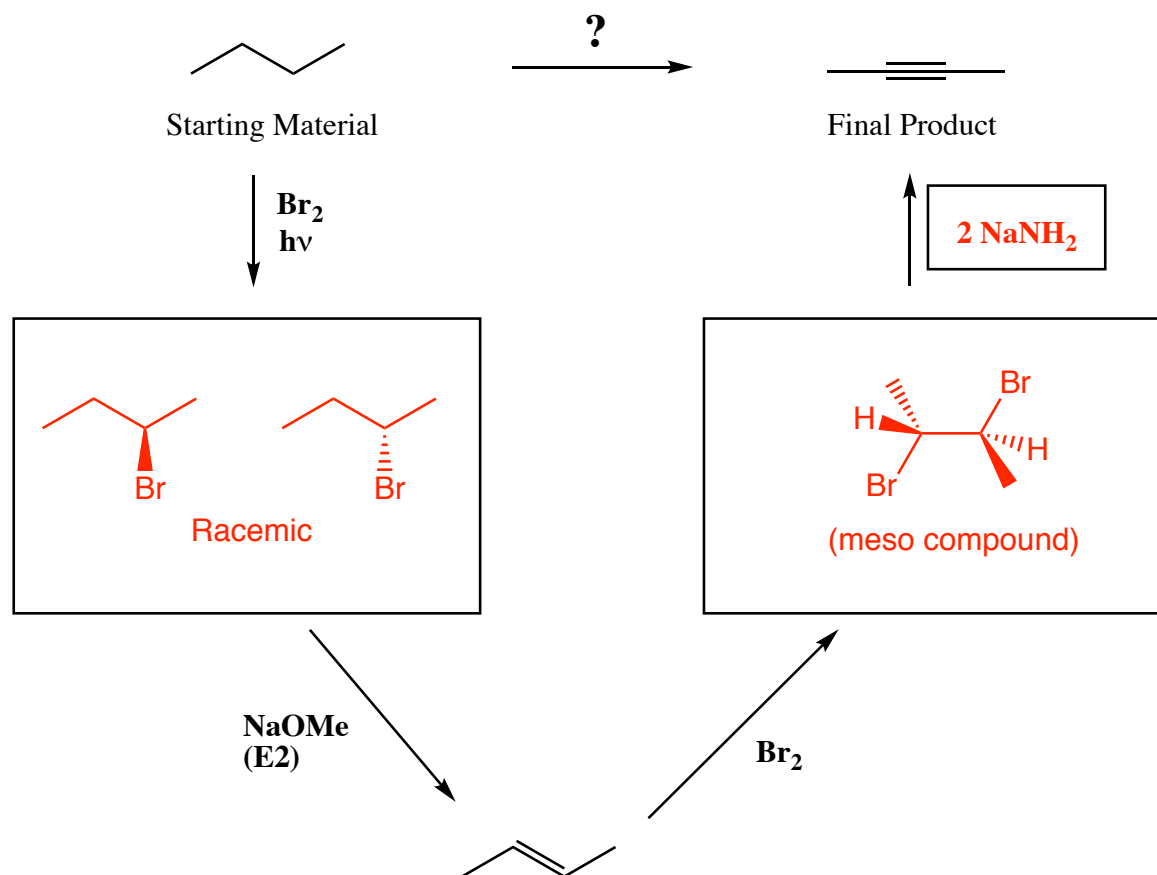
5. For the following, complete the reactions with the predominant product or products. You must indicate stereochemistry with wedges and dashes. If a racemic mixture is created, you must write "racemic" under the structures.



6. For the following, complete the reactions with the predominant product or products. You must indicate stereochemistry with wedges and dashes. If a racemic mixture is created, you must write "racemic" under the structures.



7. The point of organic chemistry is synthesis, the conversion of starting molecules into different ones with a new structure and function. Each reaction you are learning should be thought of as a “tool” that allows you to create a desired type of molecule. These tools can be used in an almost infinite number of combinations to create truly interesting molecules. **In the large boxes provided, draw the structures of the molecules indicated in this synthesis scheme. For smaller boxes next to the arrow, write the reagents needed to carry out the appropriate reaction.** You may use any reactions we have learned provided that the product(s) you draw for each step is/are the predominant one(s). Show all the reagents you need. Show each molecule synthesized along the way and be sure to pay attention to the regiochemistry and stereochemistry preferences for each reaction. You must draw all stereoisomers formed, and use wedges and dashes to indicate chirality at each chiral center. Write racemic when appropriate. **All the carbons of the product must come from carbons of the starting material.**



8. In the large boxes provided, draw the structures of the molecules indicated in this synthesis scheme. For smaller boxes next to the arrow, write the reagents needed to carry out the appropriate reaction. You may use any reactions we have learned provided that the product(s) you draw for each step is/are the predominant one(s). Show all the reagents you need. Show each molecule synthesized along the way and be sure to pay attention to the regiochemistry and stereochemistry preferences for each reaction. You must draw all stereoisomers formed, and use wedges and dashes to indicate chirality at each chiral center. Write racemic when appropriate. All the carbons of the product must come from carbons of the starting material.

