**When analyzing allylic halogenation reactions (NBS and hn)**

1. **Consider all possible allylic radicals that can be formed.**
2. **Analyze all contributing structures for all of the allylic radicals.**
3. **Add a Br atom at the site of the unpaired electron for all contributing structures for all of the allylic radicals.**
4. **From all of the possible products, the predominant product is the one THAT IS THE MOST STABLE ALKENE – the most substitued alkene – alkyl groups stabilize alkenes – *trans* over *cis*.**
5. **Note: It is OK if the product you choose derives from an allylic radical contributing structure that is a minor contributor. FOR THIS REACTION WE ONLY CARE ABOUT THE RELATIVE STABILITIES OF THE PRODUCT ALKENES.**

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