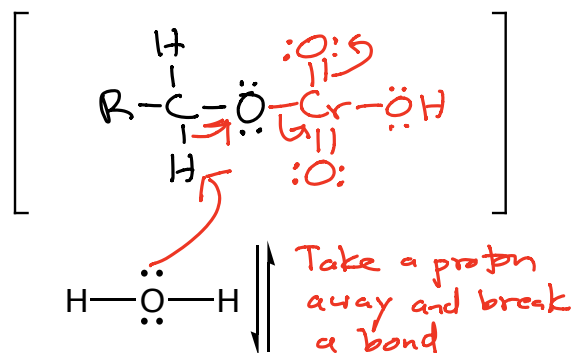
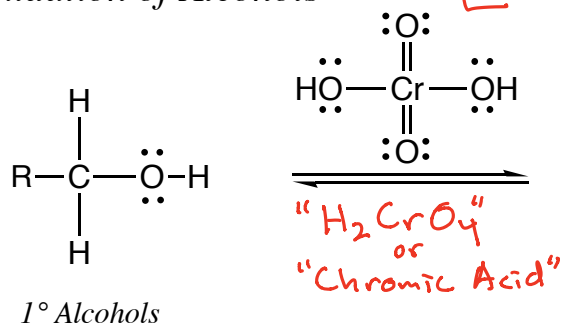


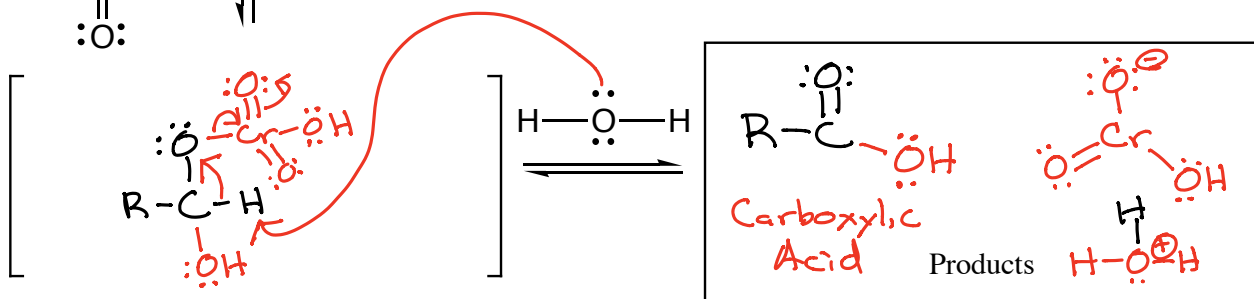
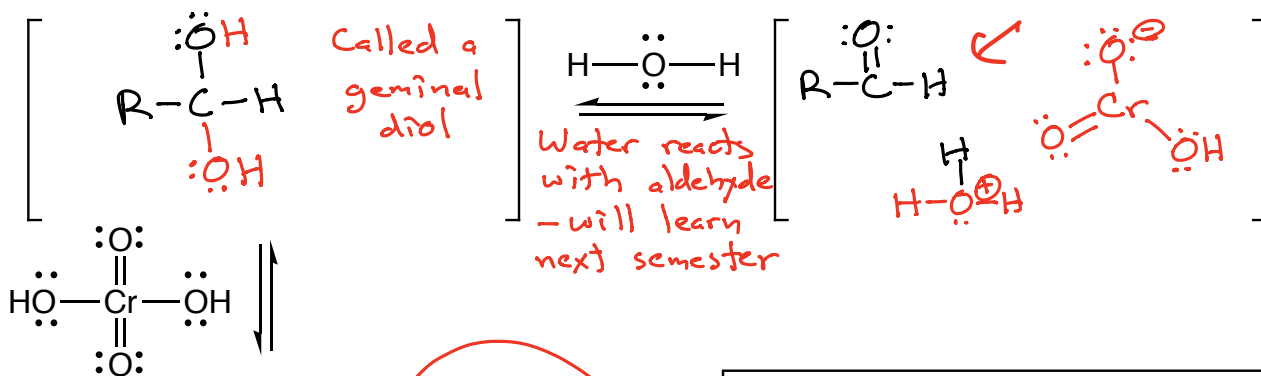
Chromic Acid Oxidation of Alcohols

Called "Jones Reagent" $\left\{ \begin{array}{l} \text{CrO}_3 + \text{H}_2\text{O} \\ \text{or} \\ \text{K}_2\text{CrO}_7 \end{array} \right\} + \text{H}_2\text{SO}_4$

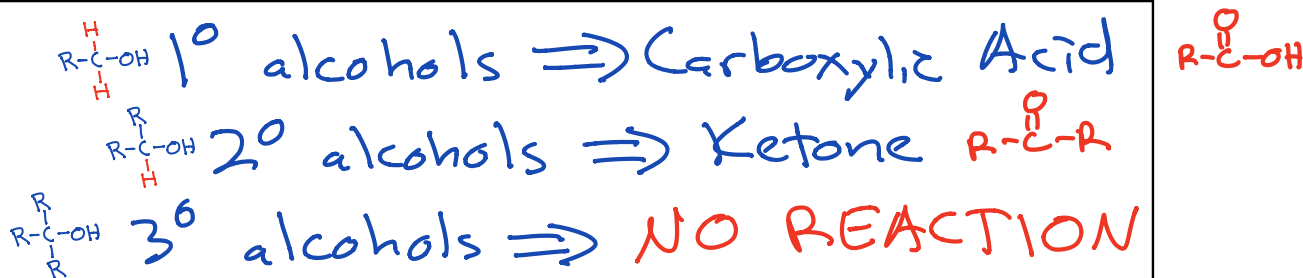
Not responsible for first step



Not responsible for this step



Summary:



Regiochemistry: N/A

Stereochemistry: N/A

Example:



A chromic acid-like reagent WITHOUT WATER will stop at the **aldehyde** when using a **primary alcohol** as starting material

