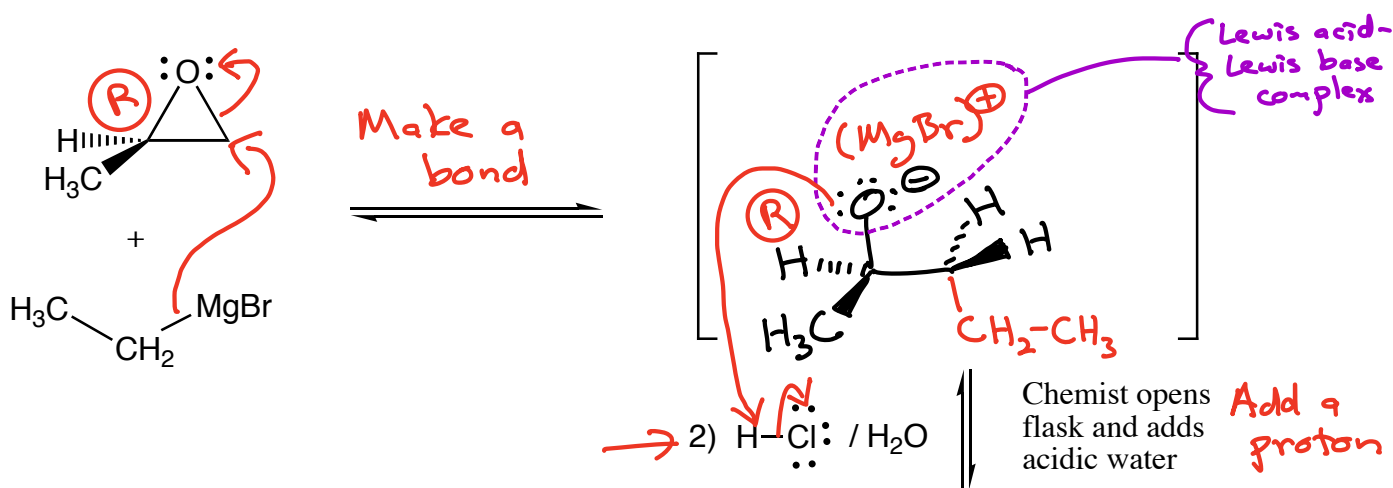


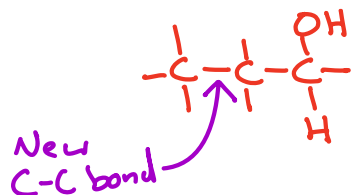
Organolithium and Gilman reagents react the same way as Grignard reagents in this reaction.

Grignard Reagent Reacting with an Epoxide

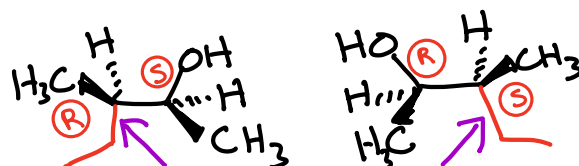
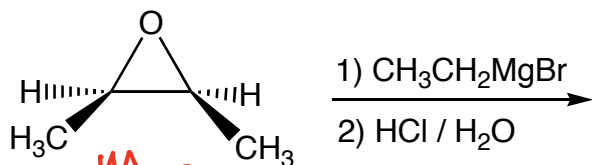
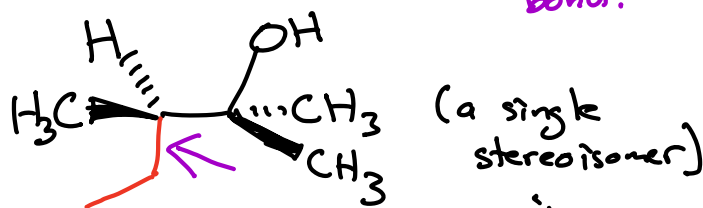
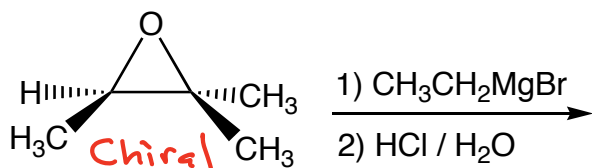
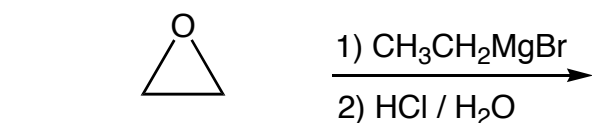
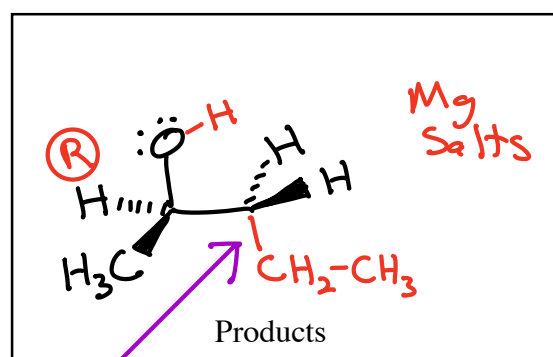


Key Recognition Element (KRE):

There is a new C-C bond that is two carbon atoms away from an OH group

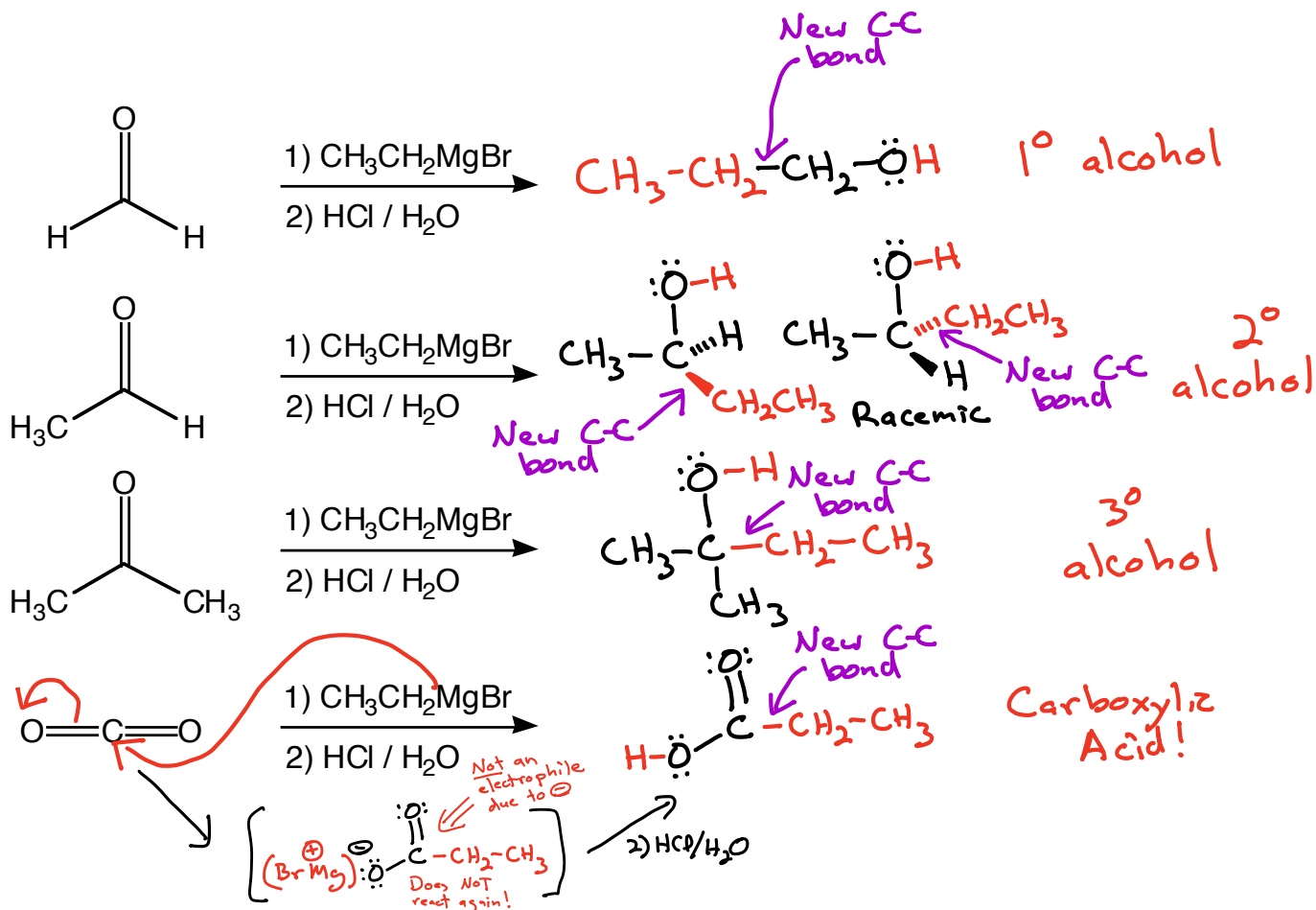
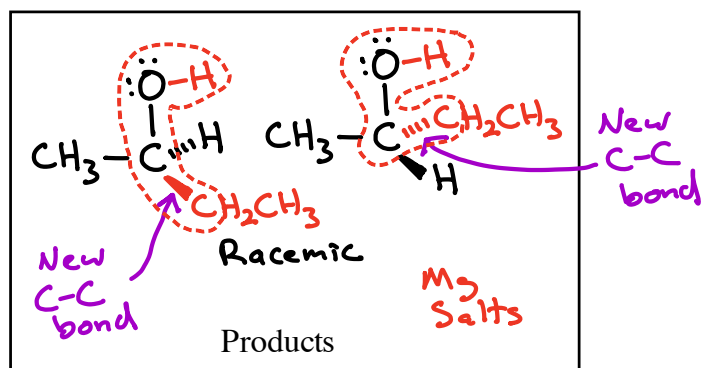
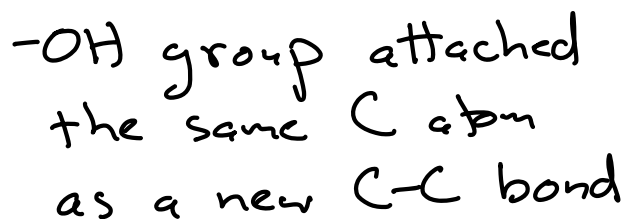


New C-C bond!

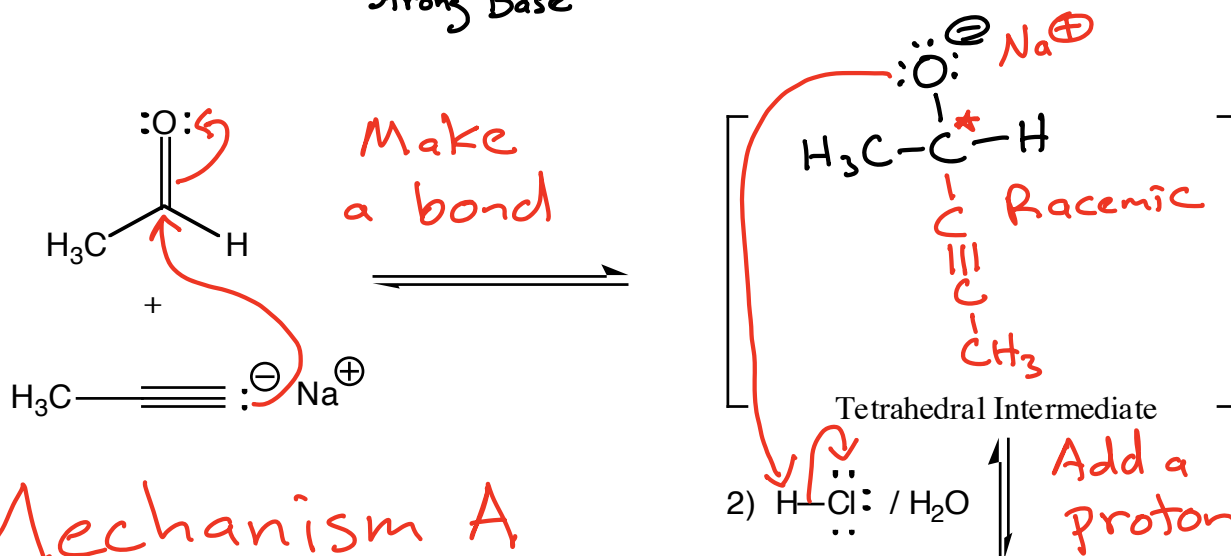
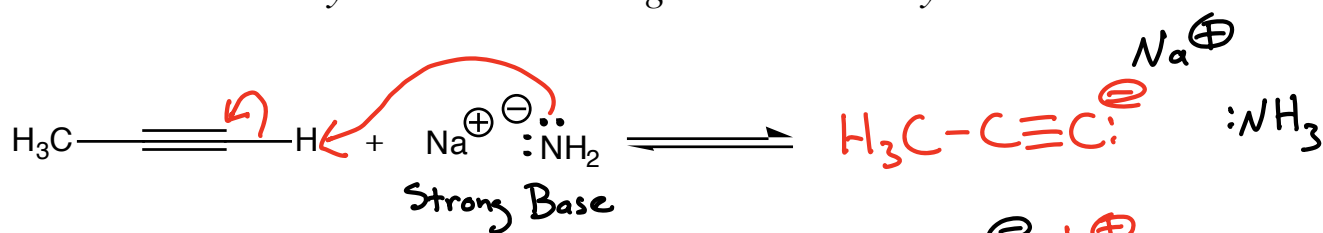


Racemic Mixture

Mechanism A



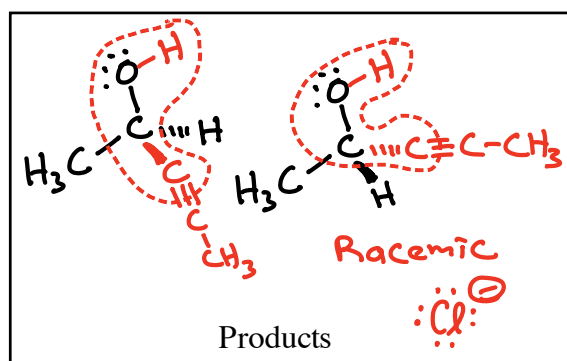
Alkyne Anion Reacting with an Aldehyde or Ketone



Mechanism A

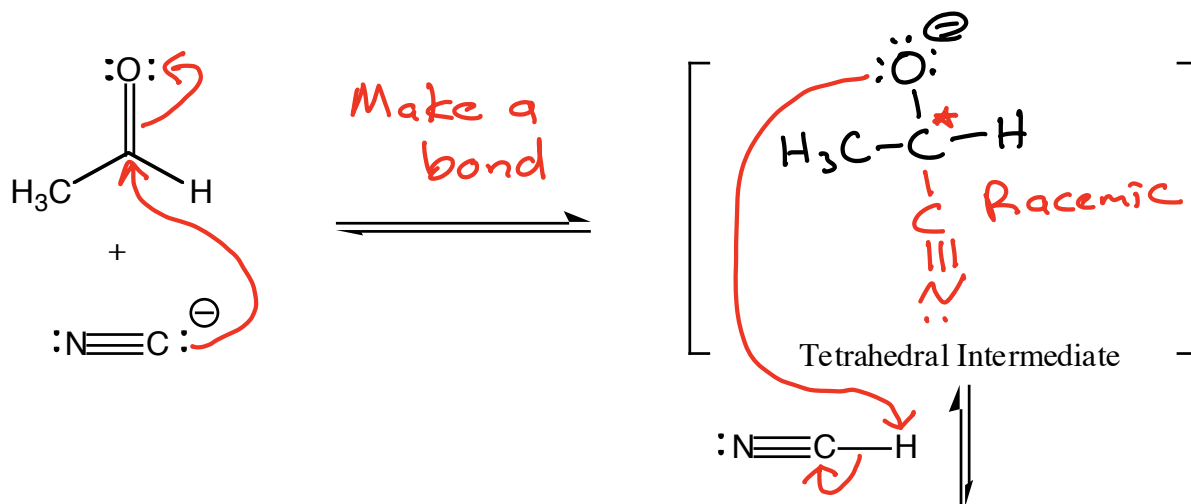
Key Recognition Element (KRE):

OH group on the carbon that makes a new C-C bond to an sp C atom (alkyne)



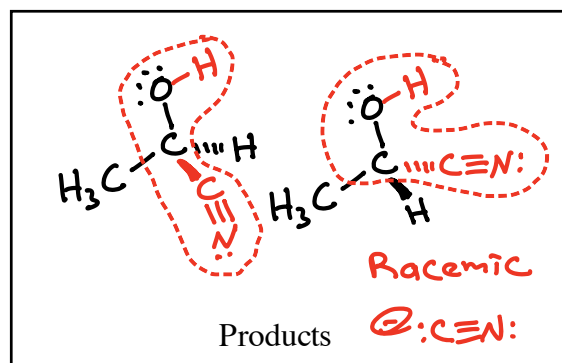
HCN Reacting with an Aldehyde or Ketone

Reacts on the C atom because that makes stronger bonds



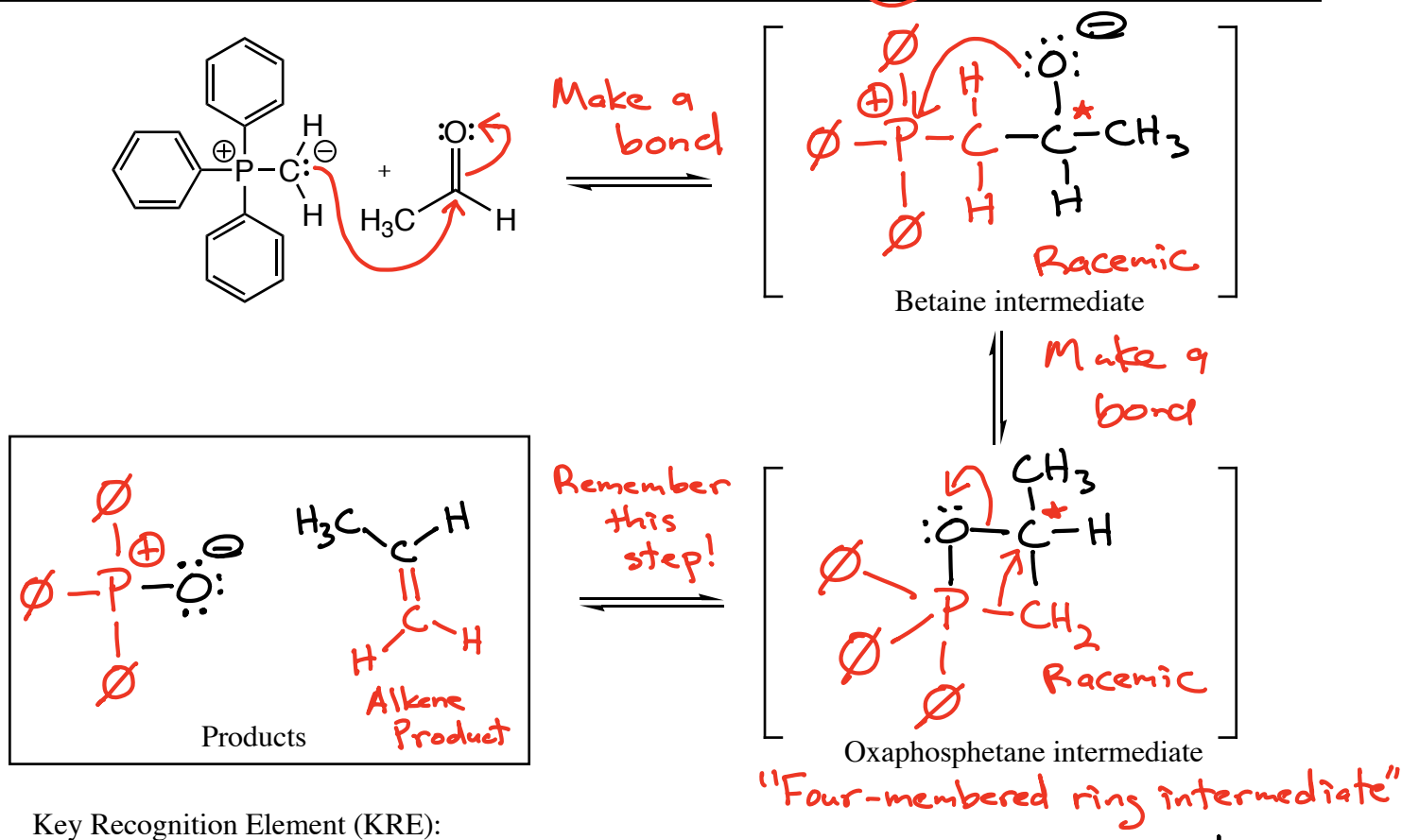
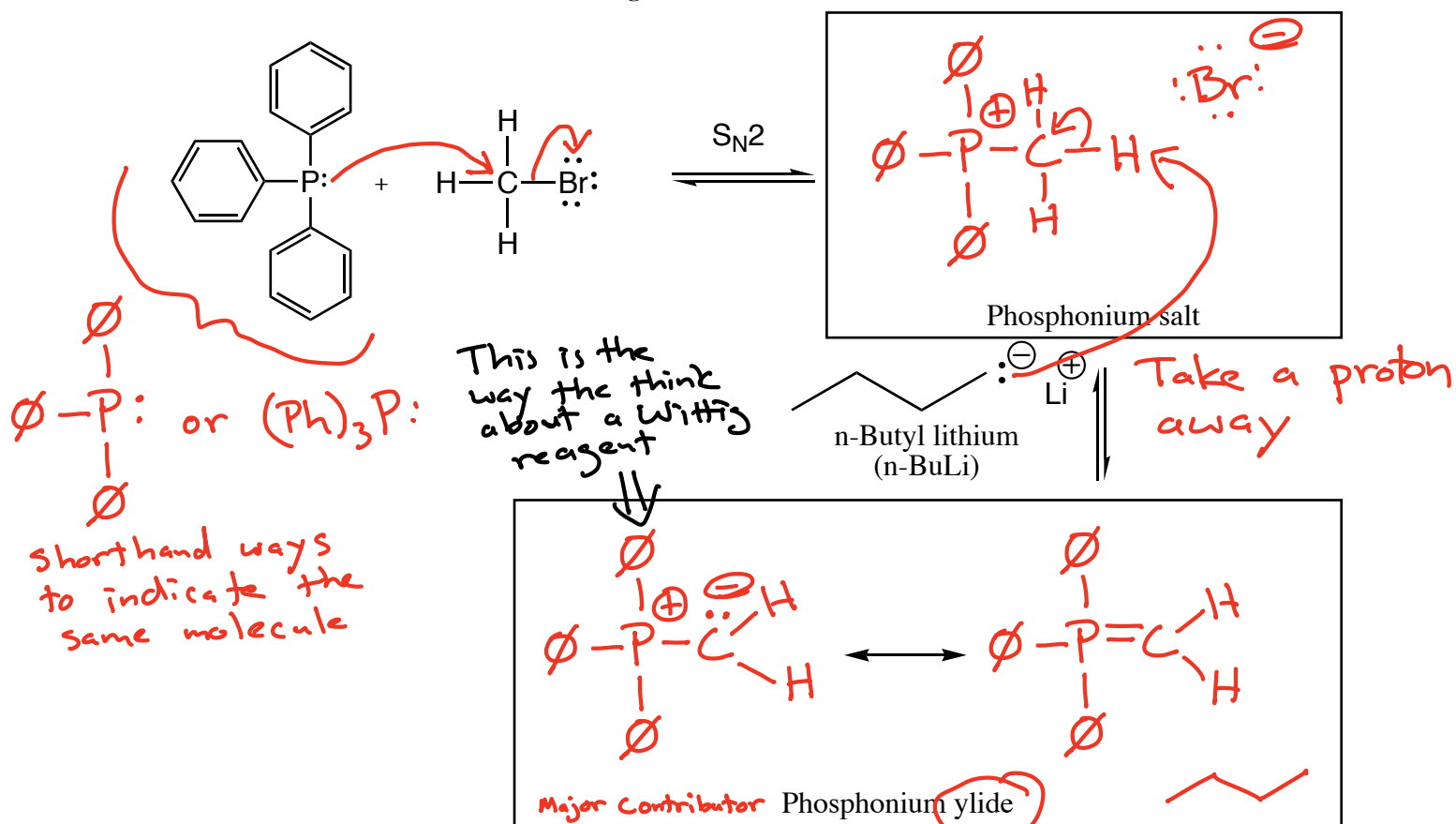
Key Recognition Element (KRE):

Cyanohydrin \rightarrow OH
on a C atom that
made a new C-C
bond to $-\text{C}\equiv\text{N:}$



Time capsule \rightarrow cyanohydrins can be
hydrolyzed in $\text{H}_2\text{SO}_4/\text{H}_2\text{O}$ to
give α -hydroxyacids
"alpha"

Wittig Reaction



Key Recognition Element (KRE):

Alkene \rightarrow New $\text{C}=\text{C}$ where the $\text{C}=\text{O}$ was!