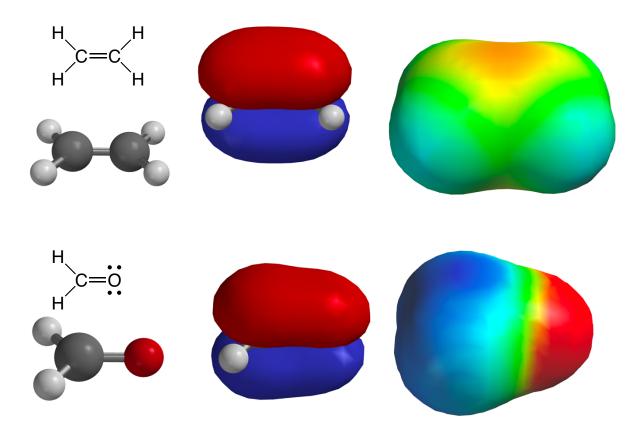
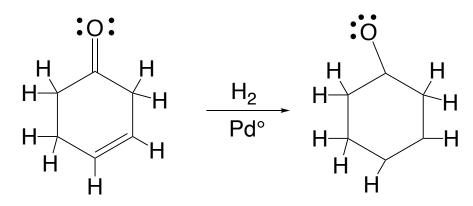
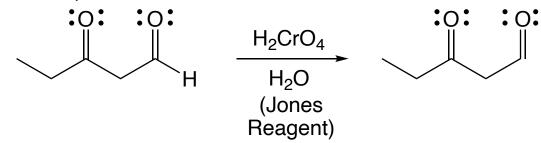
Detour: Hydrogenation and Oxidation of Aldehydes and Ketones

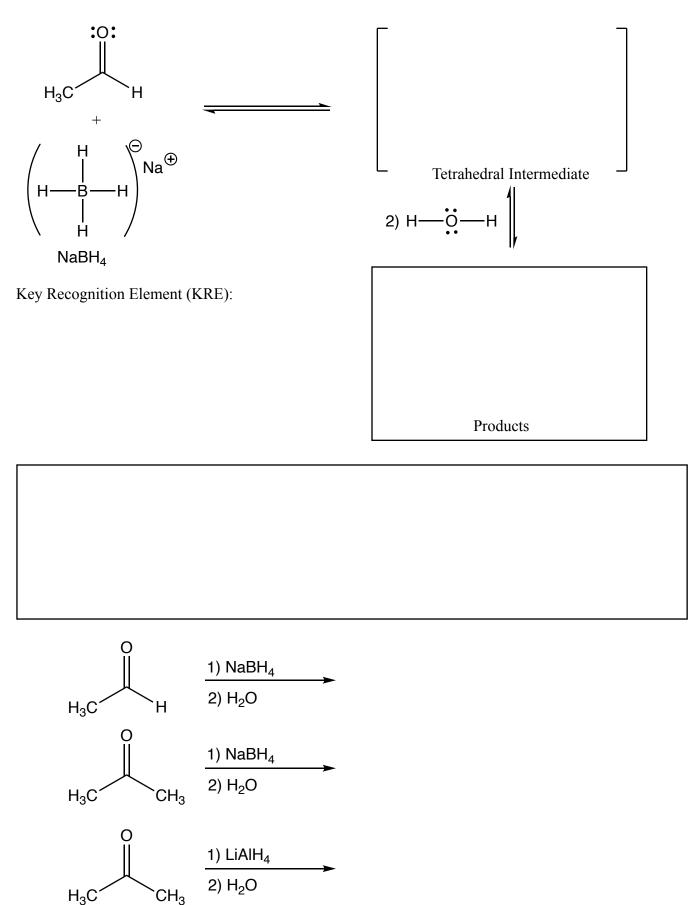


The pi bonds of carbonyls react the same as pi bonds of alkenes with  $H_2$  in the presence of Pt°, Pd° or Ni°

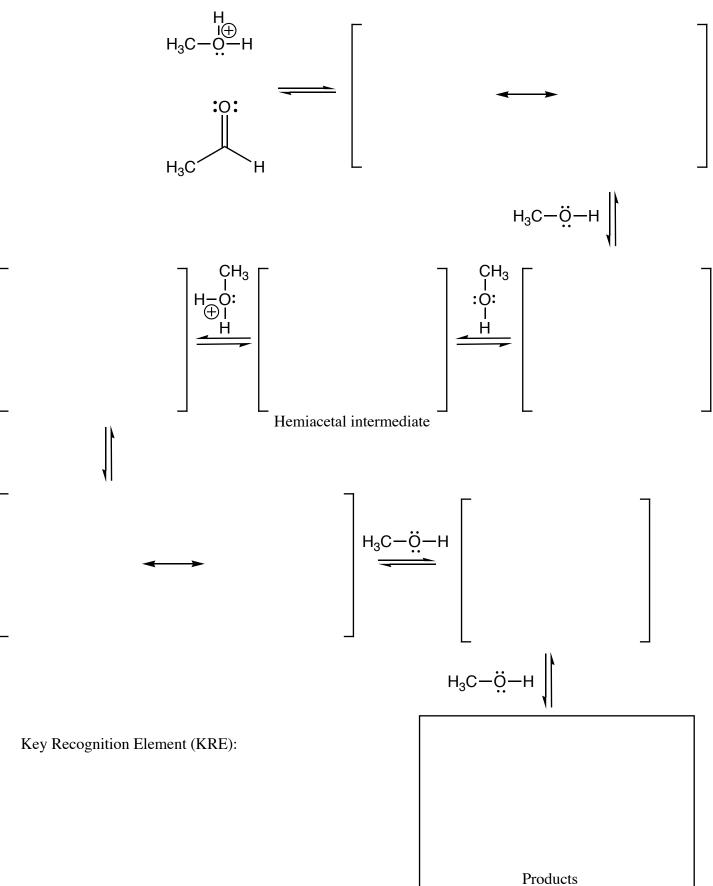


Aldehydes are oxidized to carboxylic acids using the Jones Reagent  $(H_2CrO_4 \text{ in } H_2O)$ . Ketones do not react.

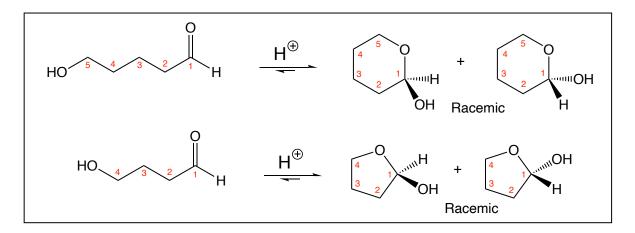


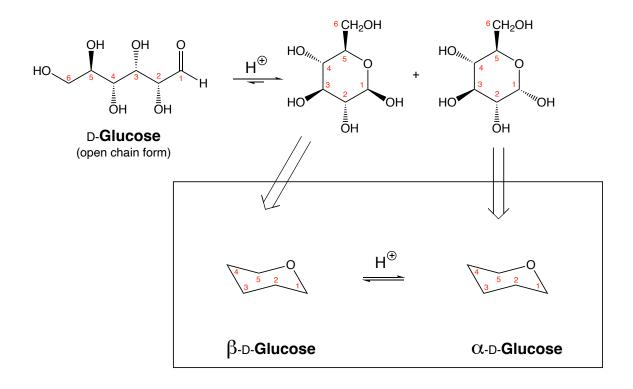


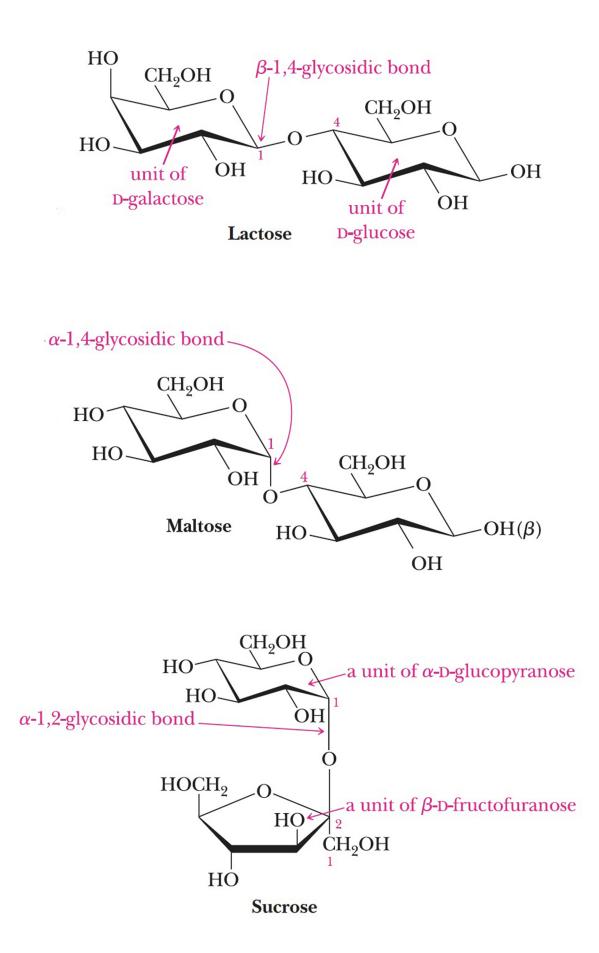
Acid Catalyzed Hemiacetal and Acetal Formation From an Aldehyde or Ketone



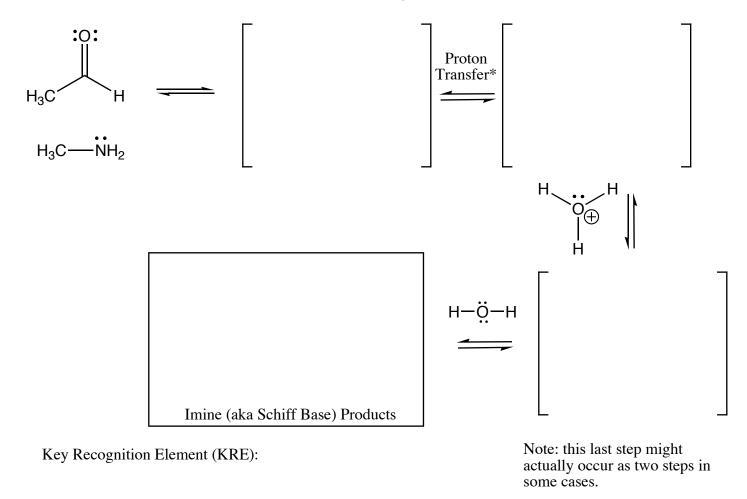
## Cyclic Hemiacetals and Carbohydrates



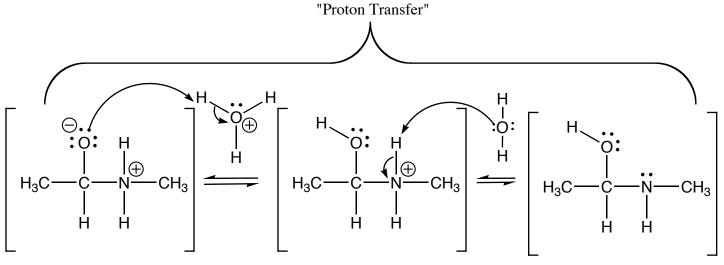




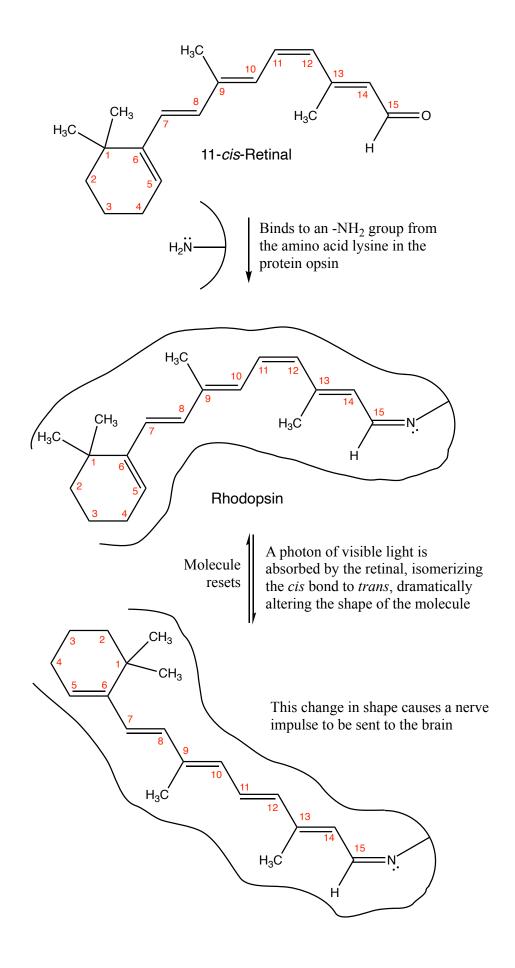
## Formation if an Imine (Schiff Base) From an Aldehyde or Ketone Reacting with an Amine



\* "Proton Transfer" refers to a situation in which a proton moves from one part of a molecule to another on the SAME MOLECULE. We do not draw arrows for proton transfer steps because that would be deceptive. In some cases, the same proton may move from one part of the molecule to the other directly, but in other cases, solvent molecules may be involved as indicated in the following scheme. To make things even more interesting, the following two steps might even be reversed in some cases. Becuase of all the ambiguity, we just write "Proton Transfer" and do not bother with arrows.



## How vision works



## Terpenes

