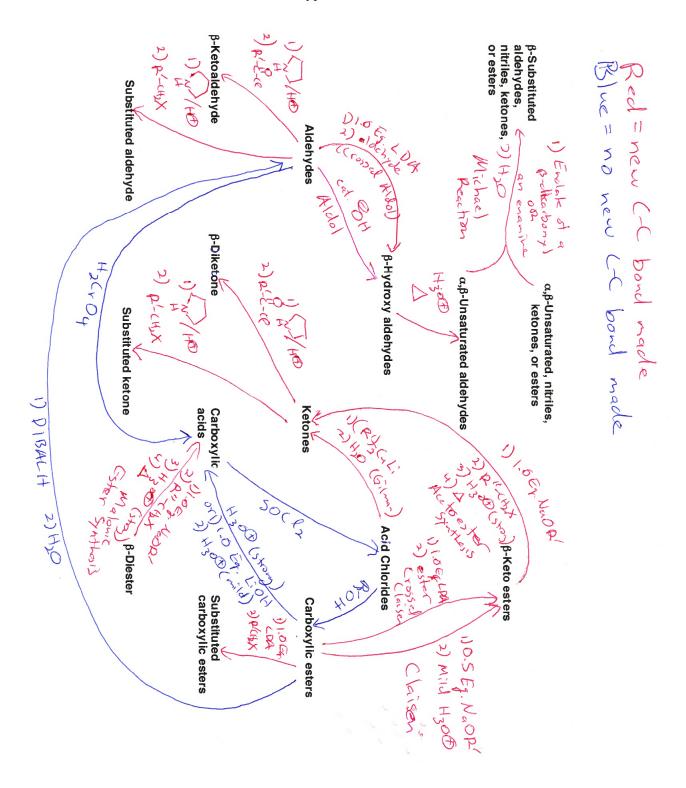
| NAME (Print): | Chemistry 320N Dr. Brent Iverson 8th Homework March 26, 2024 | |
|---|--|--|
| Please print the first three letters of your last name in the three boxes | | |

(20 Points) Fill in the missing arrows and reagents to complete this roadmap using all the reactions you have learned this semester that interconvert these types of molecules.

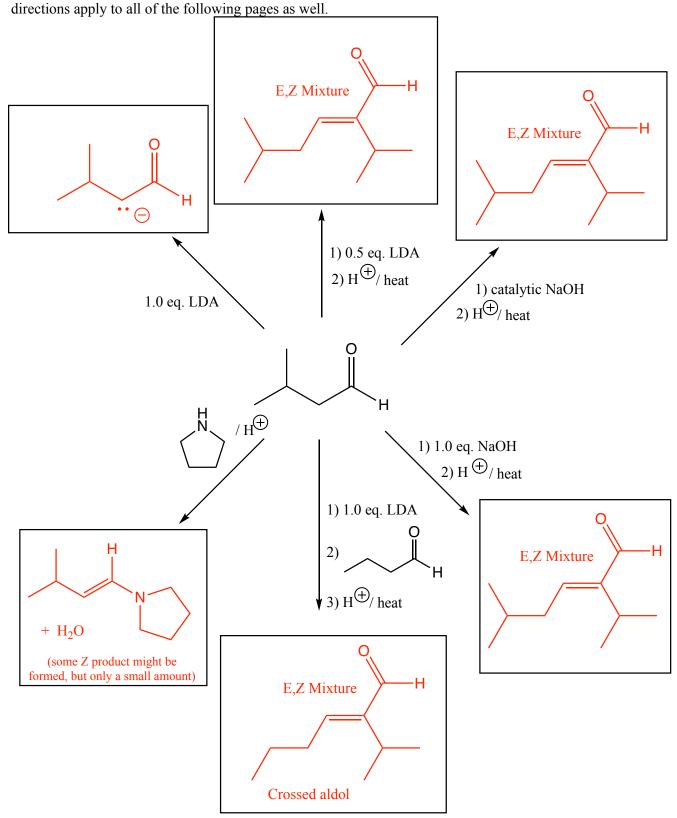


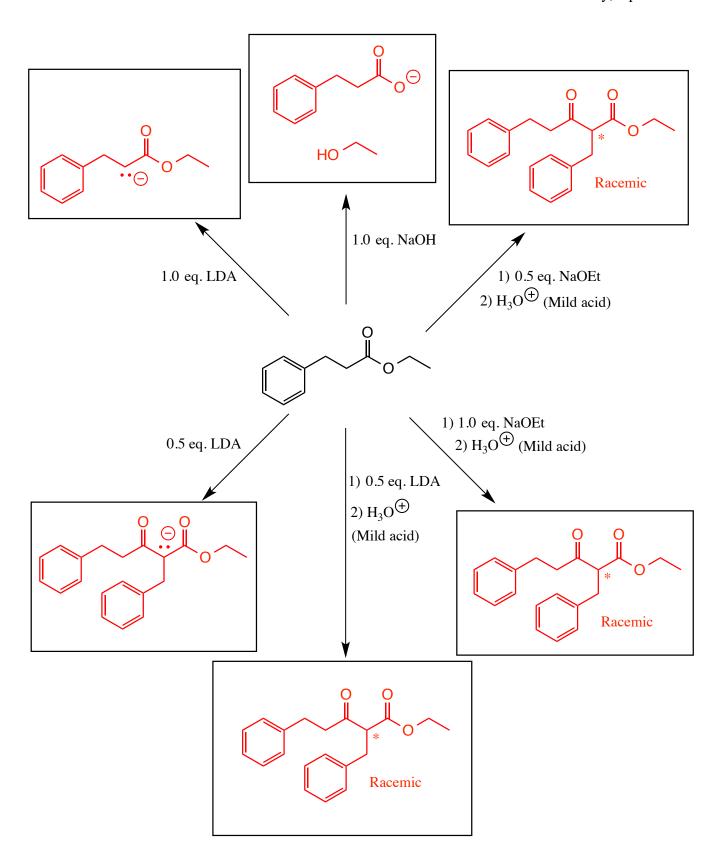
Write the predominant product that will occur for each transformation. If a new chiral center is created and a racemic mixture is formed, you must draw both enantiomers and write "racemic" under the structure. Use wedges () and dashes () to indicate stereochemistry. For these, you do not have to worry about metal salts in the products.

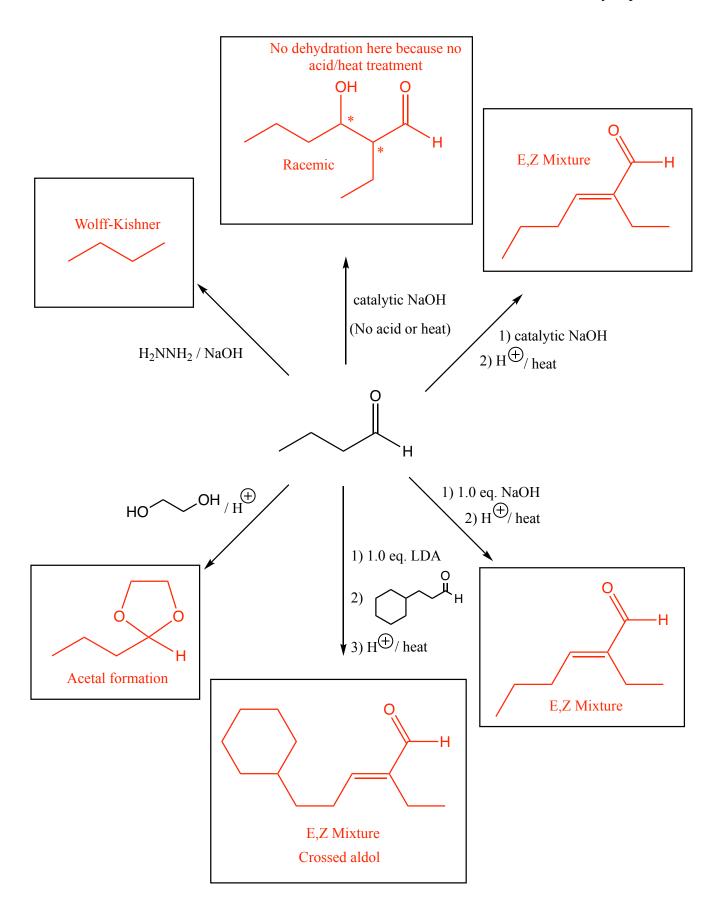
There is a lot to think about here. Please take your time. ASSUME THIS DEHYDRATES.

In each of the boxes over an arrow, write the minimum number of equivalents of the specified reagent required to carry out the reaction shown to completion. If only a catalytic amount is needed, write "CAT". Note: You must assume the carbonyl compound starting material is initially present in an amount of 1.0 equivalent. In the problems with a box, fill in the missing starting material.

(3 or 5 pts each) Fill in the boxes with the appropriate structure or structures. Because these structures are getting complex, you do not need to draw both enantiomers. Instead, when a new chiral center is created, just mark it with an asterisk (*) and label the product as "racemic". No need to use wedges and dashes. Also, when an E,Z mixture is formed, only draw one and label it as "E,Z mixture". These

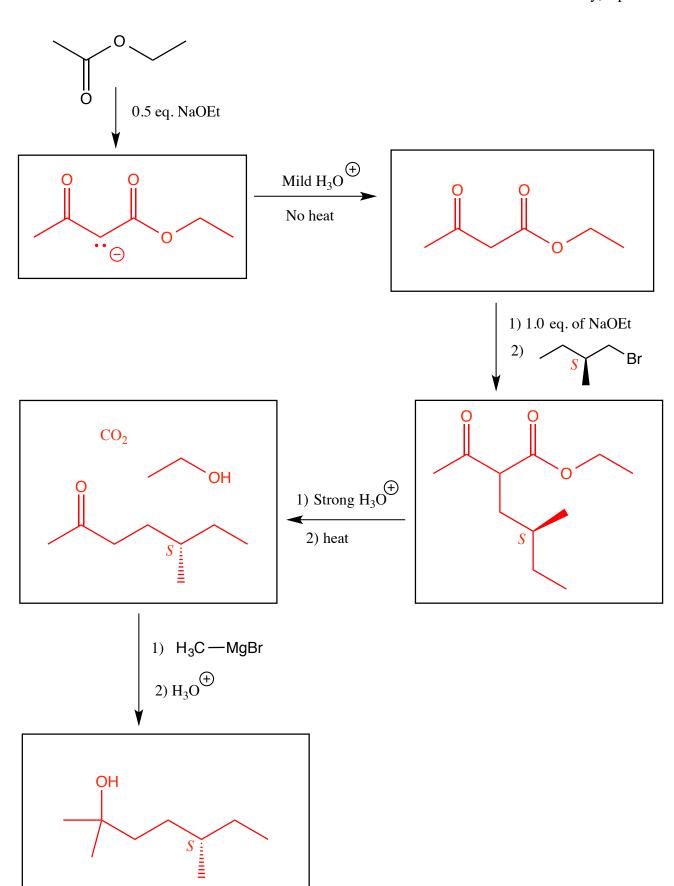






1)
$$O_3$$
2) $(CH_3)_2S$

1) catalytic NaOH
2) H^{\bigoplus} / heat



1) 1 Eq. NaOEt

3) Mild H₃O⁽⁺⁾

1) 1 Eq. NaOEt

2)

NH₂

3) Mild H₃O⁽⁺⁾