NAME (Print): _____

SIGNATURE: _____

Chemistry 320N Dr. Brent Iverson 1st Homework January 16, 2024

Please print the first three letters of your last name in the three boxes





(5 or 7 pts each) The following reactions all involve substitution and/or elimination. Fill in the box above the arrow with the mechanism that will be followed ($S_N 2$, E2, etc.). Then draw only the predominant product or products and please remember that you must draw the correct stereoisomers. For $S_N 1/E1$ reactions you must draw both types of products.







(3 pts each) Fill in the boxes with the structures that complete the reactions. Use wedges and dashes to indicate stereochemistry when appropriate. This format is intended to get you more comfortable with working backwards in synthesis problems.



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(7 and 9 pts) For the following sequences of reactions, work through all the different steps and then write the final product(s). Assume only the predominant product is formed at each step. You must indicate stereochemistry with wedges and dashes. You must draw all stereoisomers produced as predominant products and write "racemic" under the structures when appropriate. Assume no rearrangments take place.





1) Br ₂	
2) 3 NaNH ₂	
3) Mild HCl / H_2O	
4) 1 NaNH ₂	_
5)Br	

6) H₂ / Lindlar's catalyst
7) OsO₄
8) NaHSO₃

These are synthesis questions. You need to show how the starting material can be converted into the product(s) shown. You may use any reactions we have learned provided that the product(s) you draw for each step is/are the predominant one(s). Show all the reagents you need. Show each molecule synthesized along the way and be sure to pay attention to the regiochemistry and stereochemistry preferences for each reaction. You must draw all stereoisomers formed, and use wedges and dashes to indicate chirality at each chiral center. Write racemic when appropriate.

A) (8 pts)



B) (8 pts)



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