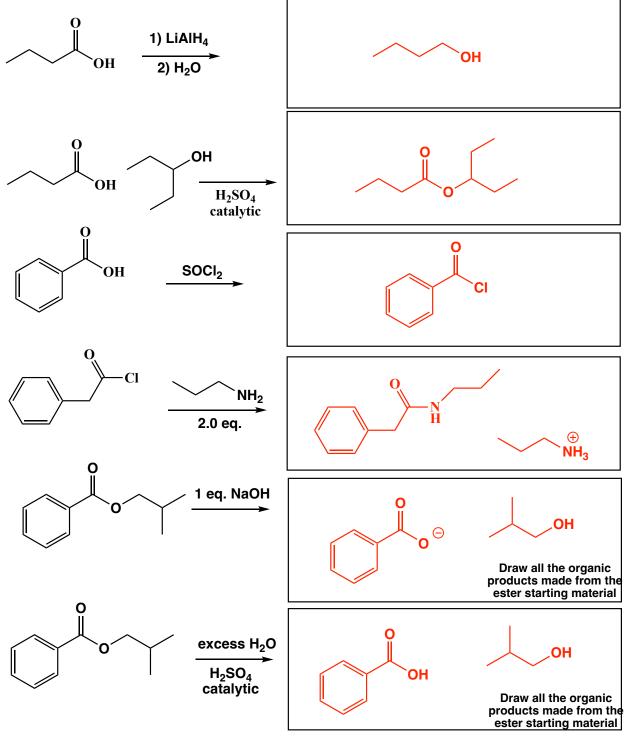
NAME (Print): _____

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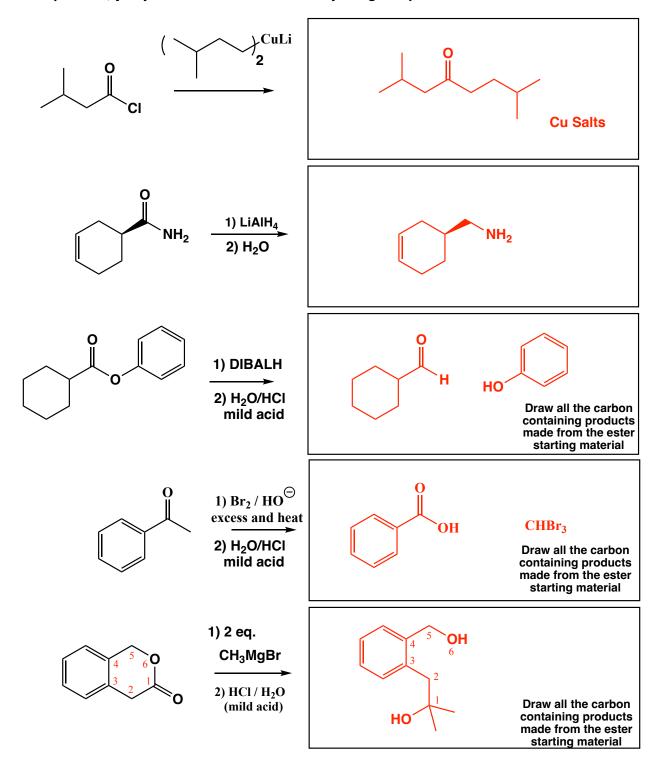
Chemistry 320N Dr. Brent Iverson **Spring Break Homework** March 7, 2024

Please print the first three letters of your last name in the three boxes

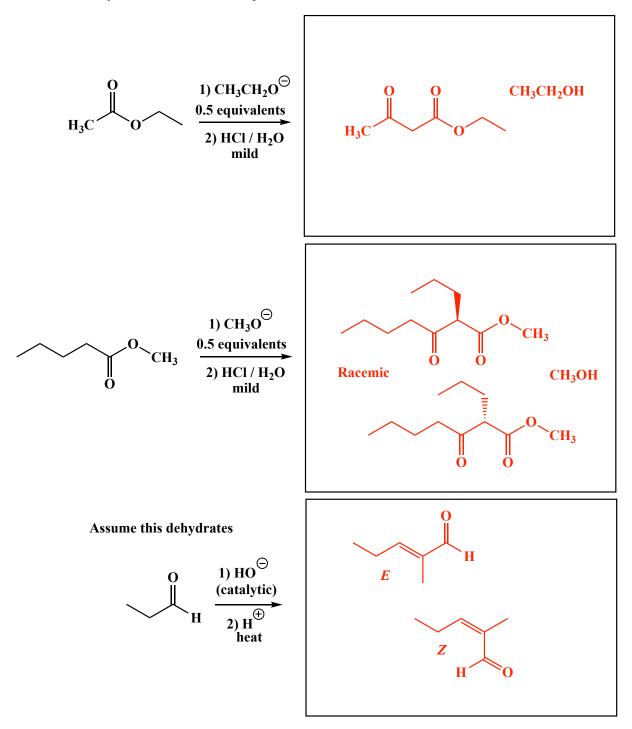
Write the predominant **carbon containing** product or products that will occur for each transformation. If there are two carbon containing products, WRITE THEM BOTH. If a new chiral center is created and a racemic mixture is formed, label the chiral center with an asterisk (*) and write racemic. No need for wedges and dashes. Also, do not worry about balancing these equations, you just need to show us the major carbon-containing products of these transformations.



Write the predominant **carbon containing** product or products that will occur for each transformation. If there are two carbon containing products, WRITE THEM BOTH. If a new chiral center is created and a racemic mixture is formed, label the chiral center with an asterisk (*) and write racemic. No need for wedges and dashes. Also, do not worry about balancing these equations, you just need to show us the major organic products of these transformations.

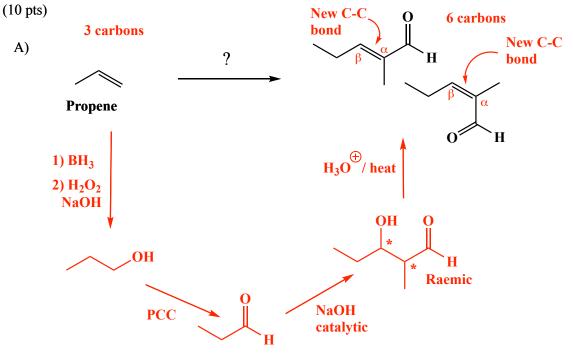


Write the predominant product that will occur for each transformation. If a new chiral center is created and a racemic mixture is formed, you must draw both enantiomers and write "racemic" under the structure. Use wedges (—) and dashes (—) to indicate stereochemistry. For these, you do not have to worry about metal salts in the products.



Using any reagents turn the starting material into the indicated product. All carbon atoms in the product must come from the starting material. Draw all molecules synthesized along the way. When in doubt, draw the molecule! Label all chiral centers with an asterisk (*) and make sure to right "Racemic" where appropriate. You will notice a theme in these problems in that you will be starting with very simple structures and making more complex products.

Remember, all of the carbons of the product must come from the given starting material.



Recognize there are 6 carbons in the product, but 3 carbons in the starting material so 2 molecules of starting material must be assembled into the product. **Recognize** further the final product as an α , β -unsaturated aldehyde, the KRE of an aldol reaction follwed by dehydration. **Recognize** the required aldehyde as being derived from the starting propene via non-Markovinikov hydroboration followed by PCC.