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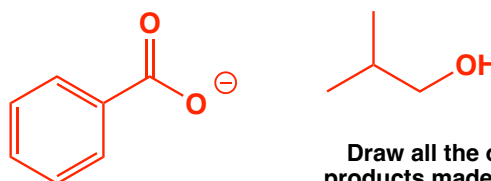
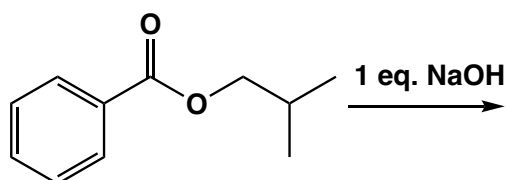
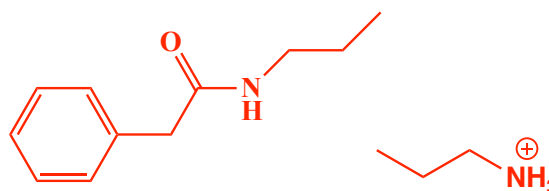
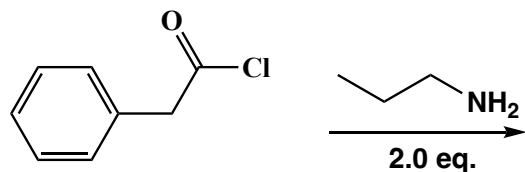
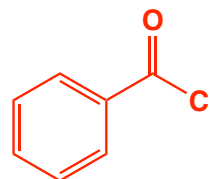
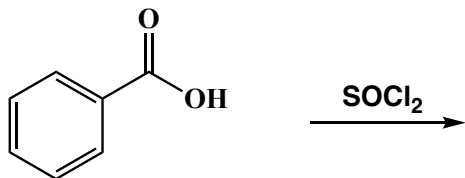
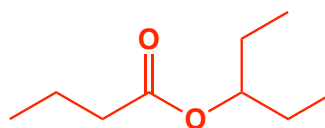
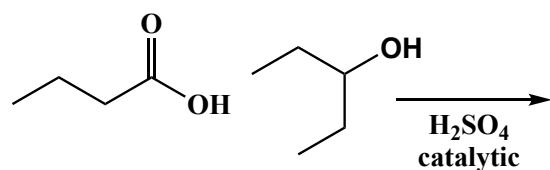
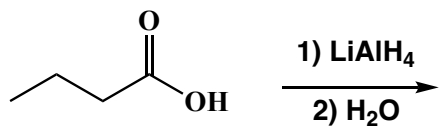
**SIGNATURE:** \_\_\_\_\_

**Chemistry 320N  
Dr. Brent Iverson  
Problem Set 7  
March 6, 2025**

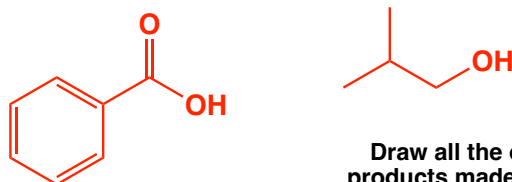
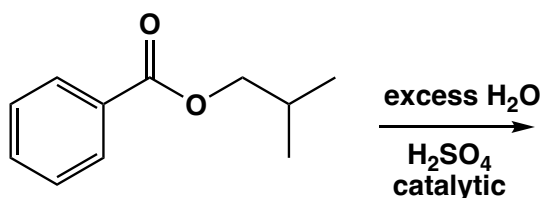
**Please print the  
first three letters  
of your last name  
in the three boxes**

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Write the predominant **carbon containing** product or products that will occur for each transformation. **If there are two carbon containing products, WRITE THEM BOTH.** If a new chiral center is created and a racemic mixture is formed, label the chiral center with an asterisk (\*) and write racemic. No need for wedges and dashes. Also, do not worry about balancing these equations, you just need to show us the major carbon-containing products of these transformations.

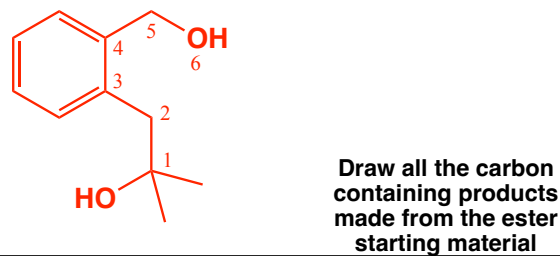
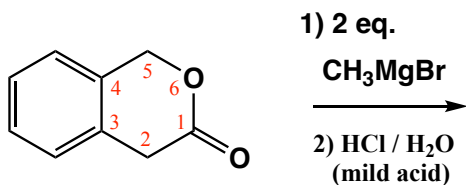
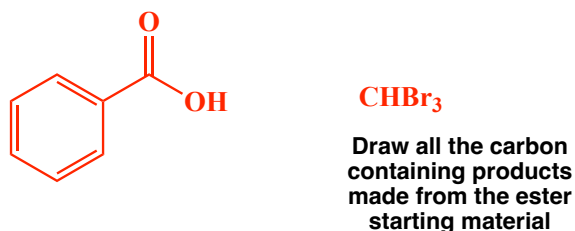
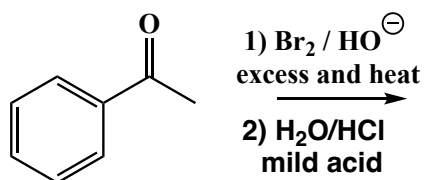
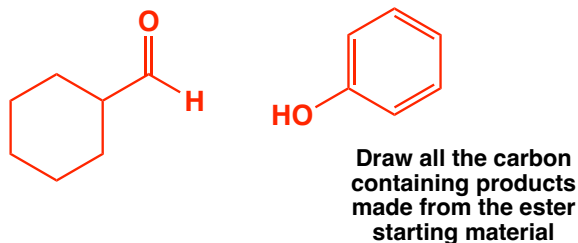
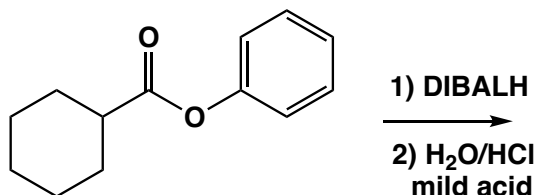
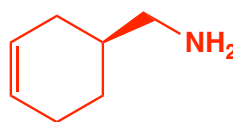
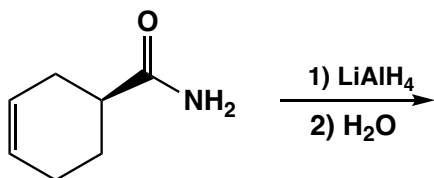
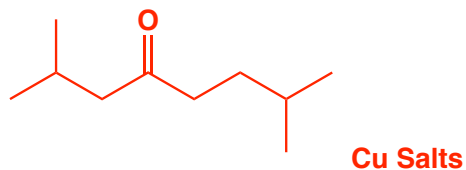
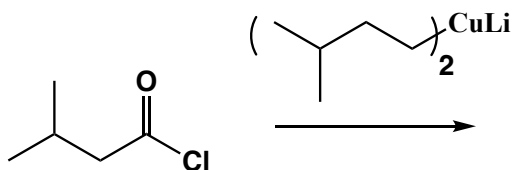


Draw all the organic products made from the ester starting material

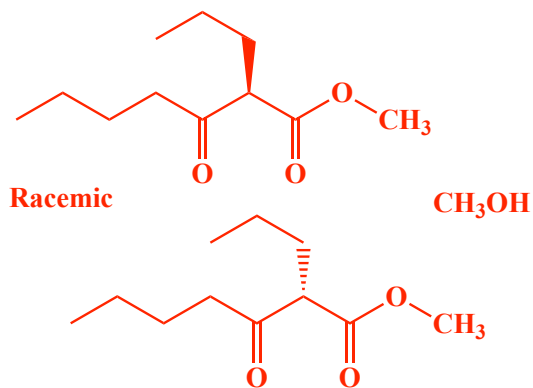
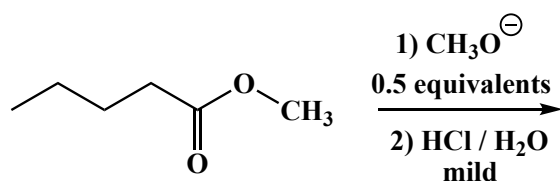
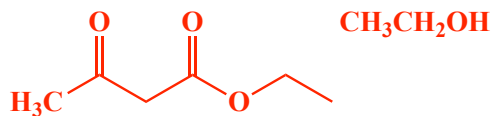
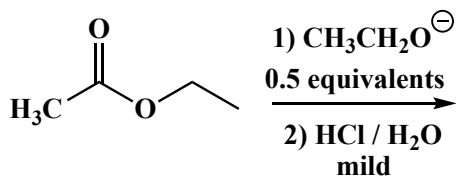


Draw all the organic products made from the ester starting material

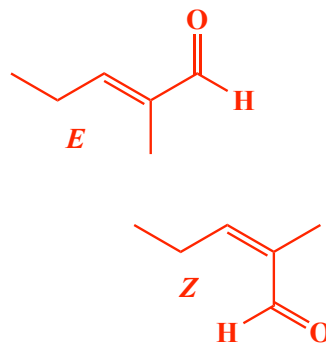
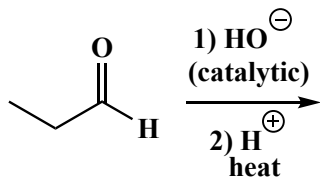
Write the predominant **carbon containing** product or products that will occur for each transformation. **If there are two carbon containing products, WRITE THEM BOTH.** If a new chiral center is created and a racemic mixture is formed, label the chiral center with an asterisk (\*) and write racemic. No need for wedges and dashes. Also, do not worry about balancing these equations, you just need to show us the major organic products of these transformations.



Write the predominant product that will occur for each transformation. If a new chiral center is created and a racemic mixture is formed, you must draw both enantiomers and write "racemic" under the structure. Use wedges (  $\blacktriangleleft$  ) and dashes (  $\cdots$  ) to indicate stereochemistry. For these, you do not have to worry about metal salts in the products.



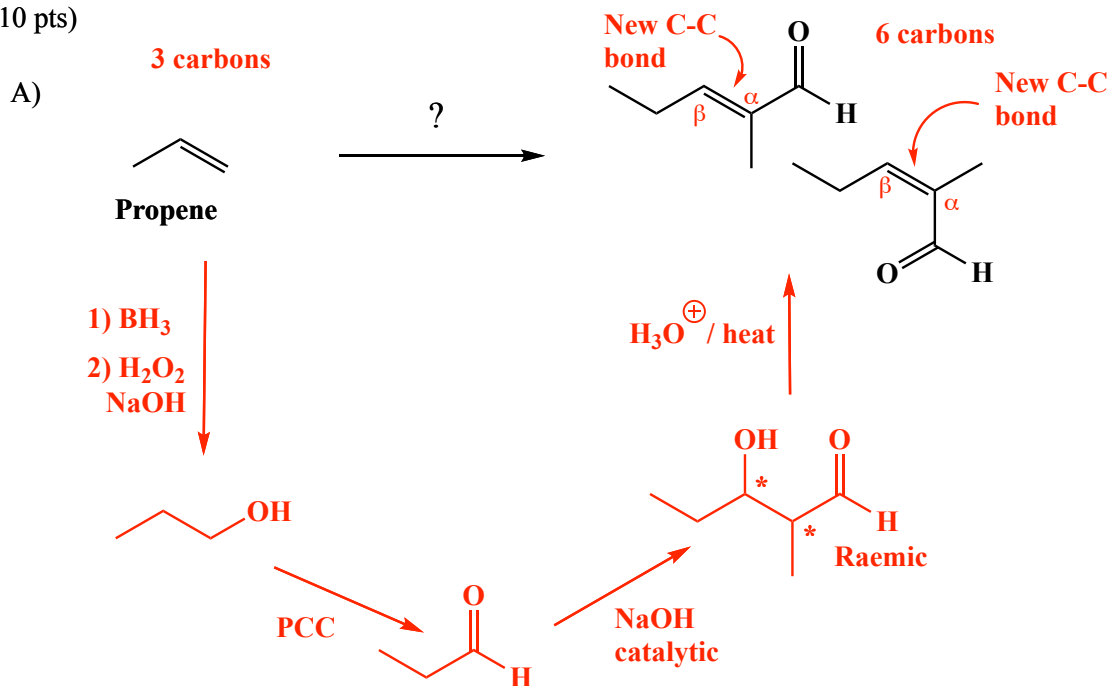
Assume this dehydrates



Using any reagents turn the starting material into the indicated product. All carbon atoms in the product must come from the starting material. Draw all molecules synthesized along the way. When in doubt, draw the molecule! Label all chiral centers with an asterisk (\*) and make sure to right "Racemic" where appropriate. You will notice a theme in these problems in that you will be starting with very simple structures and making more complex products.

Remember, all of the carbons of the product must come from the given starting material.

(10 pts)



**Recognize** there are 6 carbons in the product, but 3 carbons in the starting material so 2 molecules of starting material must be assembled into the product. **Recognize** further the final product as an  $\alpha,\beta$ -unsaturated aldehyde, the KRE of an aldol reaction followed by dehydration. **Recognize** the required aldehyde as being derived from the starting propene via non-Markovnikov hydroboration followed by PCC.