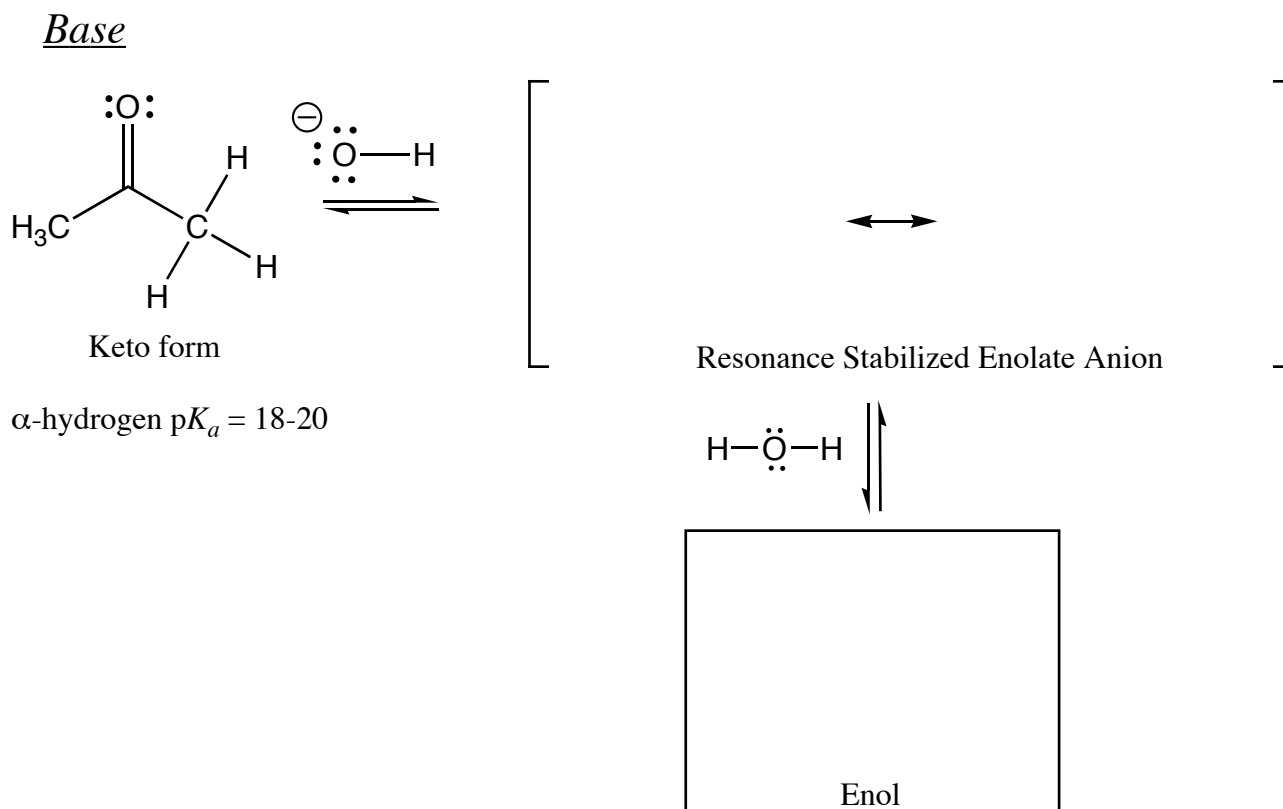
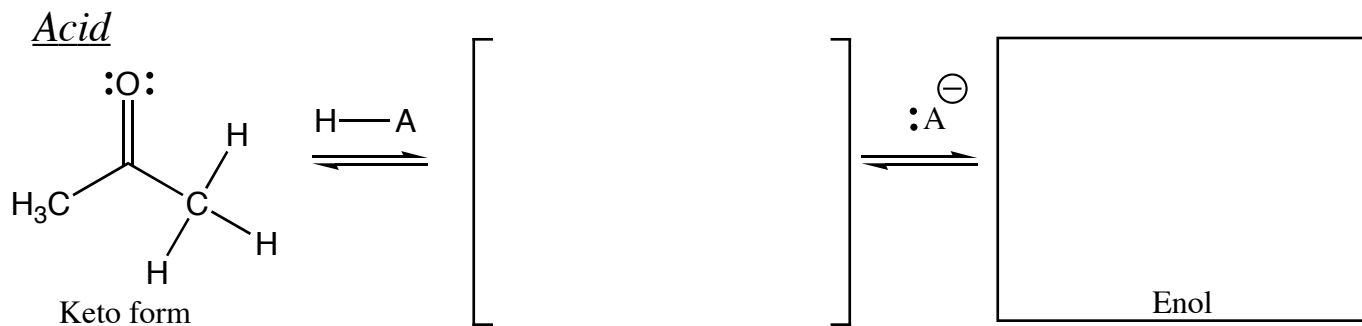


Keto-Enol Equilibrium Catalyzed by Acid or Base



For both aldehydes and ketones, the keto form predominates at equilibrium, because _____ bonds are stronger than _____ bonds.

Enols are significant, however, because they react like _____, not carbonyls, and this is important in certain situations.